

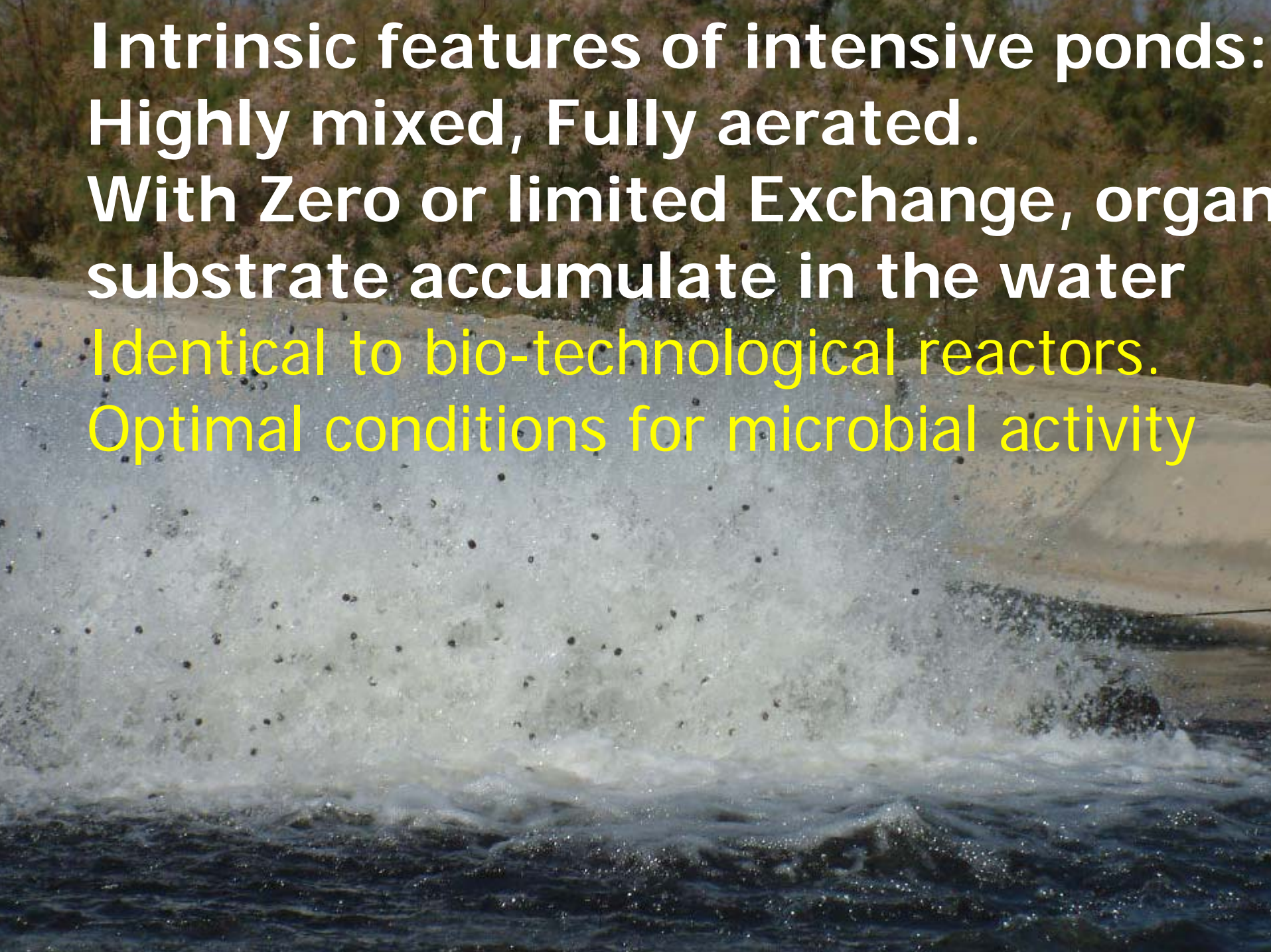
A large number of orange fish, likely tilapia, are swimming in a tank of water. The fish are densely packed, and their movement creates ripples and splashes on the water's surface. The water is a murky, brownish-green color.

Bio floc Technology Industry Session

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Of Technology**

1. Limited exchange ponds and build up Of microbial community.
2. Control of ammonia
3. Bio flocs. Feeding on bio flocs





Intrinsic features of intensive ponds:
Highly mixed, Fully aerated.
With Zero or limited Exchange, organic
substrate accumulate in the water
Identical to bio-technological reactors.
Optimal conditions for microbial activity

Characteristics of water:

- 1. Generous supply of feed to microbes
- A. Assume 3 kg fish/m³, 15 g feed/day
~ 7.5 g C
- B. ~ 50% released to the water ~ 3.8 g/m³* day
- C. On a steady state basis, with $\varepsilon = 50\%$, daily added microbial biomass carbon = 1.9g
- Average cell volume = 0.7 μm^3
- Average carbon/cell = $5.6 * 10^{-16}$ g
- Computed number of bacteria produced per day = $4.8 * 10^9$ /ml



Characteristics of water:

2



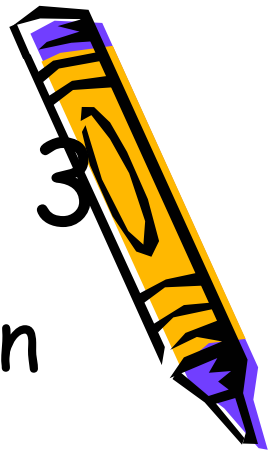
- D. Turbid water
- E. Organic carbon in water in the order of 100 mg/l
- F. Number of bacteria counts in ponds
- Around $10^7 - 10^9$ /ml
 - (Compare with computed production of 10^{10} /ml, We see high turn over rate and young microbial population)



Characteristics of water: 3

- Rate of organic matter degradation in ponds and tanks ~ **0.15 /day** (Avnimelech et al., in tanks and 113 commercial ponds samplings) to **0.27/day** in laboratory experiments (Torres Beristain 2005), as compared to **0.1-0.2** in waste water treatment plants.

i.e: **10-20% of the organic matter degrade daily.**



**Conclusions:
BFT systems are actually
Very efficient
bio-technological plants**



The nitrogen syndrome

- Fish use just about 25% of feed nitrogen. The rest excreted. 4 kg of feed protein are needed to produce 1 kg fish protein!!
- Excretion and microbial mineralization generate ammonium. **Ammonia is highly toxic.**
- Nitrite is also toxic, especially in fresh water systems.
- Ponds are enriched in N as compared to C. Carbon is emitted as CO₂. Nitrogen is left in the pond. **Can we revert this feature??**



Two Problems

- 1. Elevated inorganic N levels in the water is often the limiting factor toward high performance of intensive systems. **Inorganic nitrogen concentrations have to be controlled.**
- 2. Low efficiency of protein utilization is a waste of money (**Protein is the most expensive feed component**). In addition, since major source of protein is fish meal, harvested from over exploited oceans, **This is also a major environmental issue.**



Microbial "nutrition"

* Bacteria are made of protein

Assume feeding with sugar:

$\text{CHOH} \rightarrow \text{CO}_2 + \text{H}_2\text{O} +$
energy (~50%)

The rest \rightarrow Microbial growth

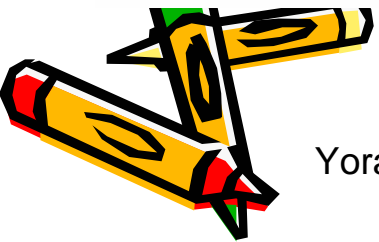
Yet, for growth they need nitrogen that is taken from the water!!






My Diet is Eating
Just Sugar so I stay
slim

I Got fat, I eat
sugar + NH₄ from
the water.



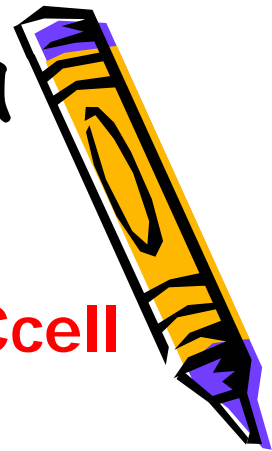
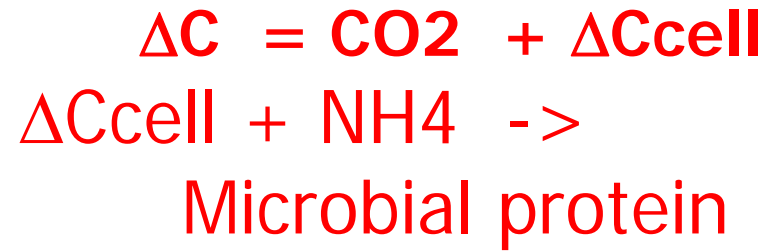
A photograph of an aeration tank in a wastewater treatment plant. The tank is filled with water, and several mechanical aerators are visible, including a large blue one and several orange ones. The aerators are actively mixing the water, creating a large amount of white foam and splashing. The background shows the concrete structure of the tank.

**We can control NH_4 Accumulation
By adding carbo-hydrates
(Sugar, mollasses, cassawa etc.**

**BACTERIA WILL TAKE UP NH_4
AND CONVERT IT INTO
MICROBIAL PROTEIN**

Manipulating bacteria

- We can add carbon rich and protein poor material (carbohydrate, CH), such as starch or cellulose (ground grains, molasses, cassava etc.) To induce accelerated nitrogen uptake.



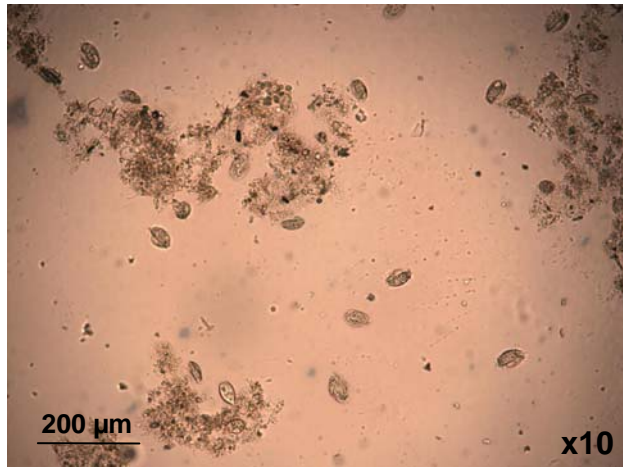
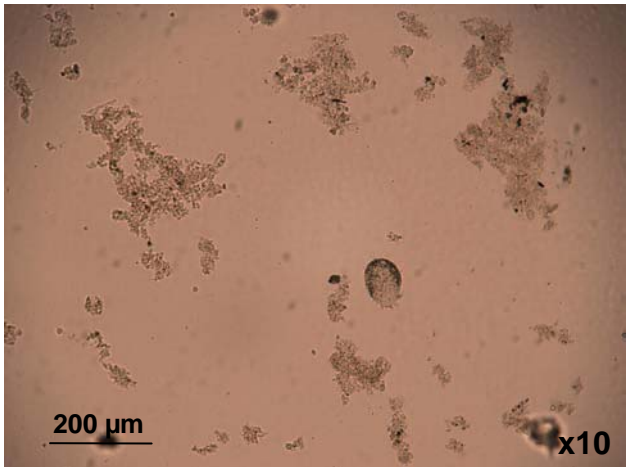
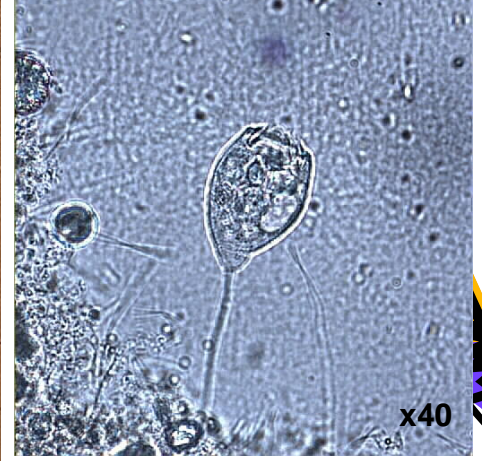
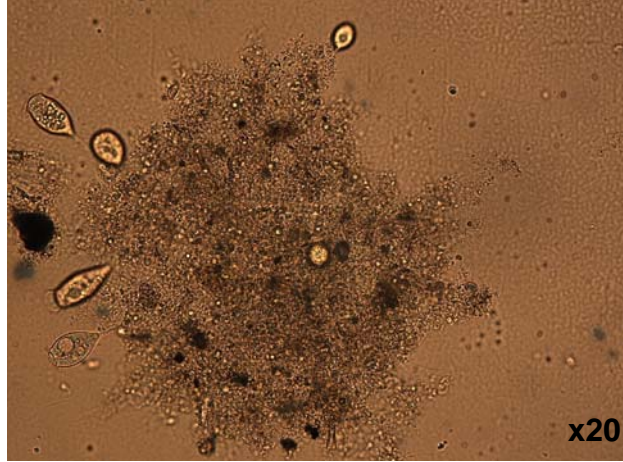
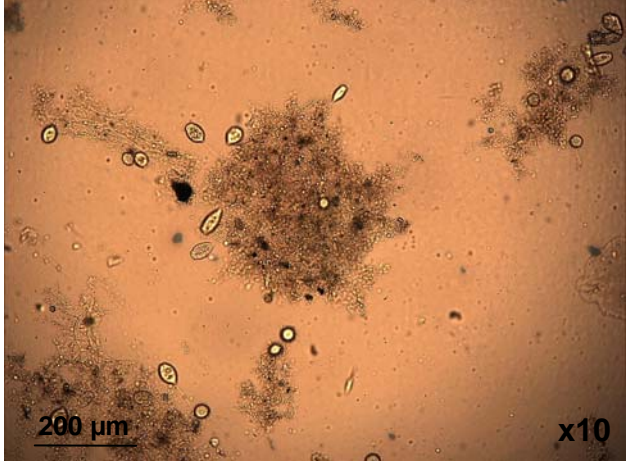
*Inorganic nitrogen control
is achievable
and predictable*



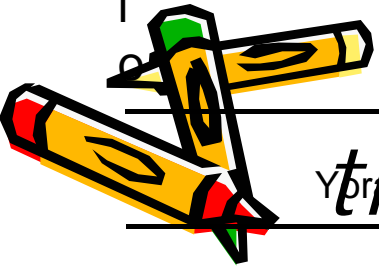
Feeding fish with bacteria

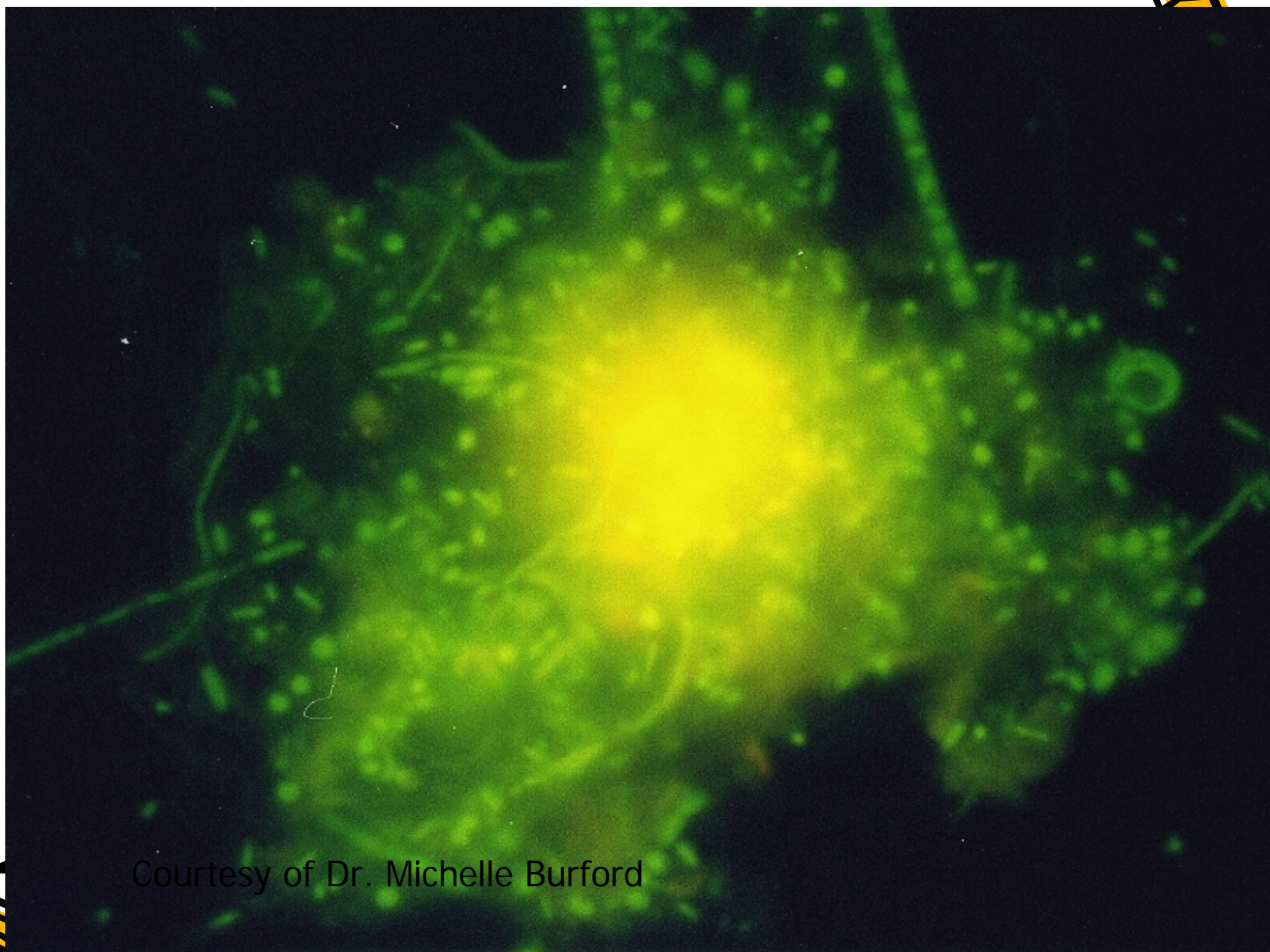
- We can induce the production of microbial protein. Will it be a good feed source for fish???
 - Can they physically harvest bacteria? **Individual bacteria are too small ($\sim 1\mu\text{m}$)**
 - Is it nutritive? **Bacterial proteins are different.**
- Will they digest it? **Probably so**



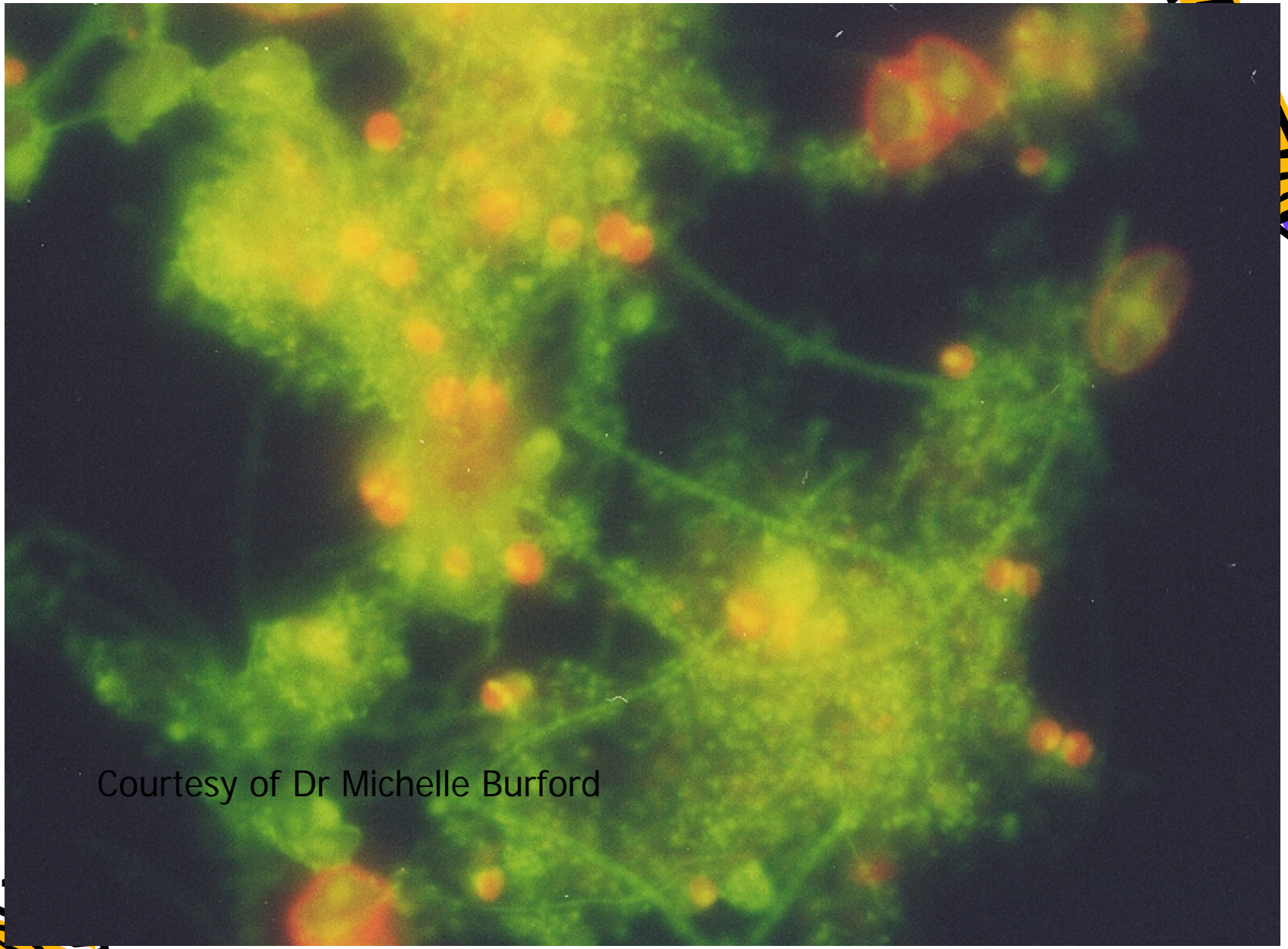


Bio flocs are made of bacteria, protozoa, etc. Typically their diameter is 0.1-2 mm.



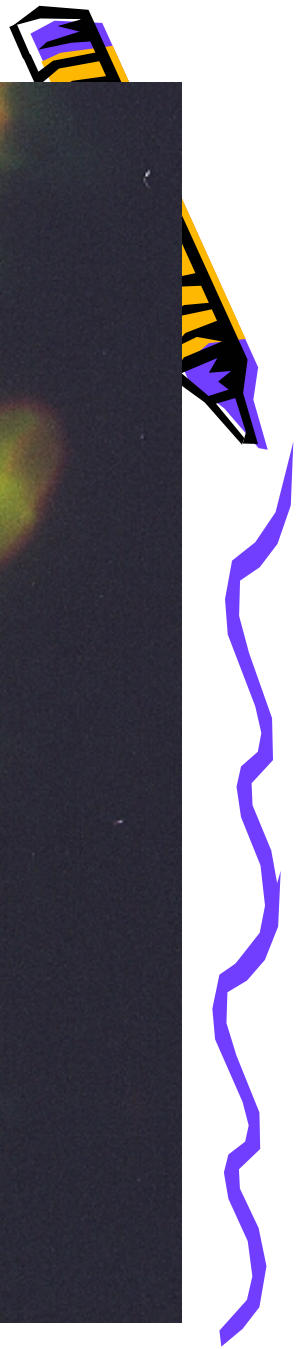


Courtesy of Dr. Michelle Burford



Courtesy of Dr Michelle Burford

Yoram Avnimelech



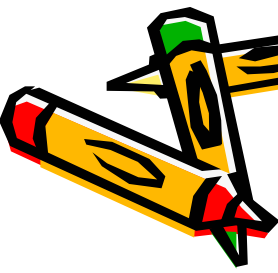
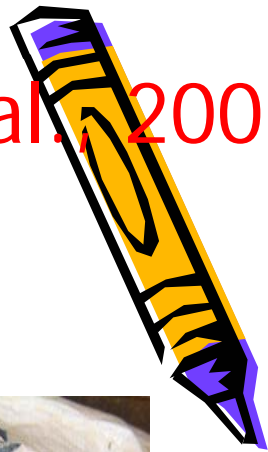


Results of a study in Belize (Burford et al. 2003)

The proportion of daily nitrogen requirement of shrimp contributed by natural biota (present mostly as bio flocs) was found, using ^{15}N

uptake study, to be

18-29%.



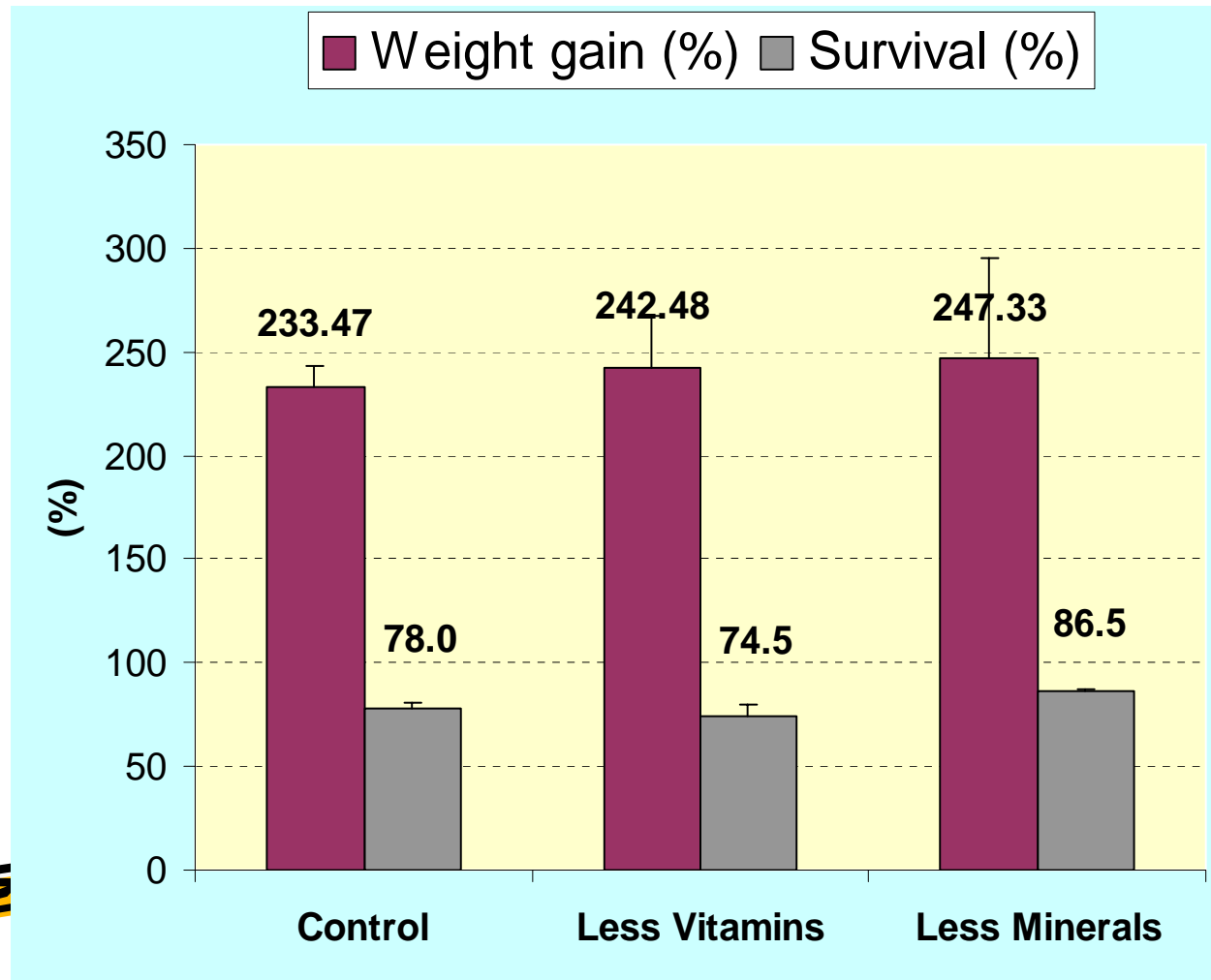
Protein Recycling

- Normally, fish or shrimp recover just ~25% of feed protein.
- In bacterial controlled ponds, they eat the protein twice; Once in the feed and then they consume microbial protein. The protein recovery in experimental and commercial systems reaches almost 50%. An efficiency of 63% was achieved in experimental tanks (Velasco et al., 1998)

Protein is the most expensive part of the feed!! •



Results of Albert Tacon (2002)



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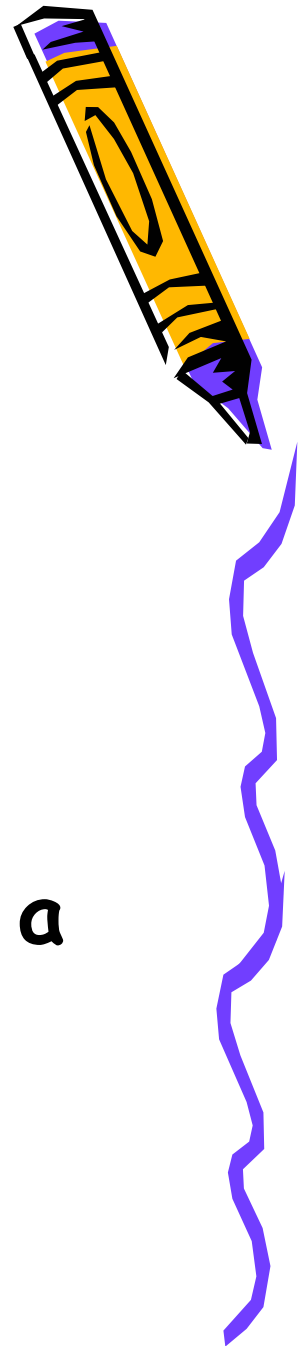
<u>Expt. # 1</u> 51 days	<u>30% Protein</u>	<u>20% Protein</u>
FEED C/N	11.1	16.6
Daily Gain (%)	1.59^a	2.0^b
FCR	2.62	2.17
PCR	4.38	2.42
<u>(Kg fish/US\$)FEED COST</u>	0.848	0.583
Exp. # 2 (30 days)		
<u>C/N</u>	11.1	16.6
Daily gain (%)	1.63^a	2.22^b
FCR	2.62	2.02
PCR	4.35	2.18
Feed cost (US\$/Kg fish)	0.848	0.543

Pros & Cons of microbial N recycling



- 1. Effective, reliable and predictable inorganic nitrogen control.
- 2. Double protein utilization, thus enables to use cheaper feed: lower protein feed.
- 3. Lower aquaculture dependence on marine fish meal & oil.



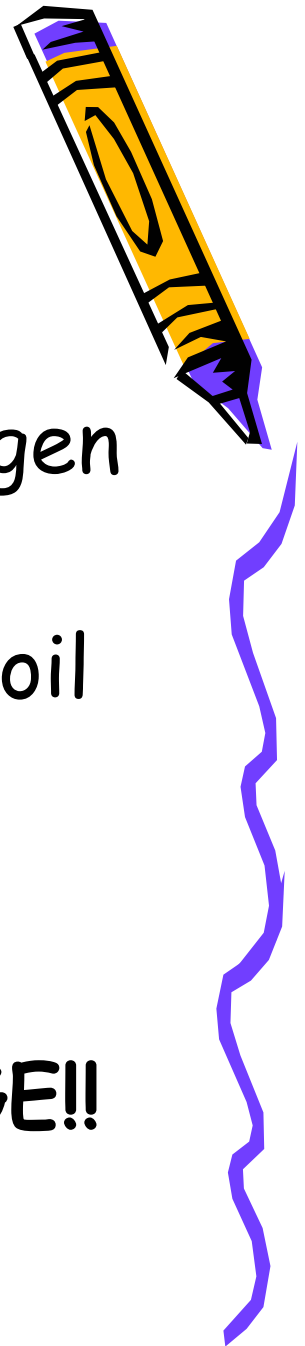


- 4. lower pollution
- 5. Slightly higher oxygen consumption.
- 6. High water turbidity, may be a problem to some species.
- 7. Probiotic effects

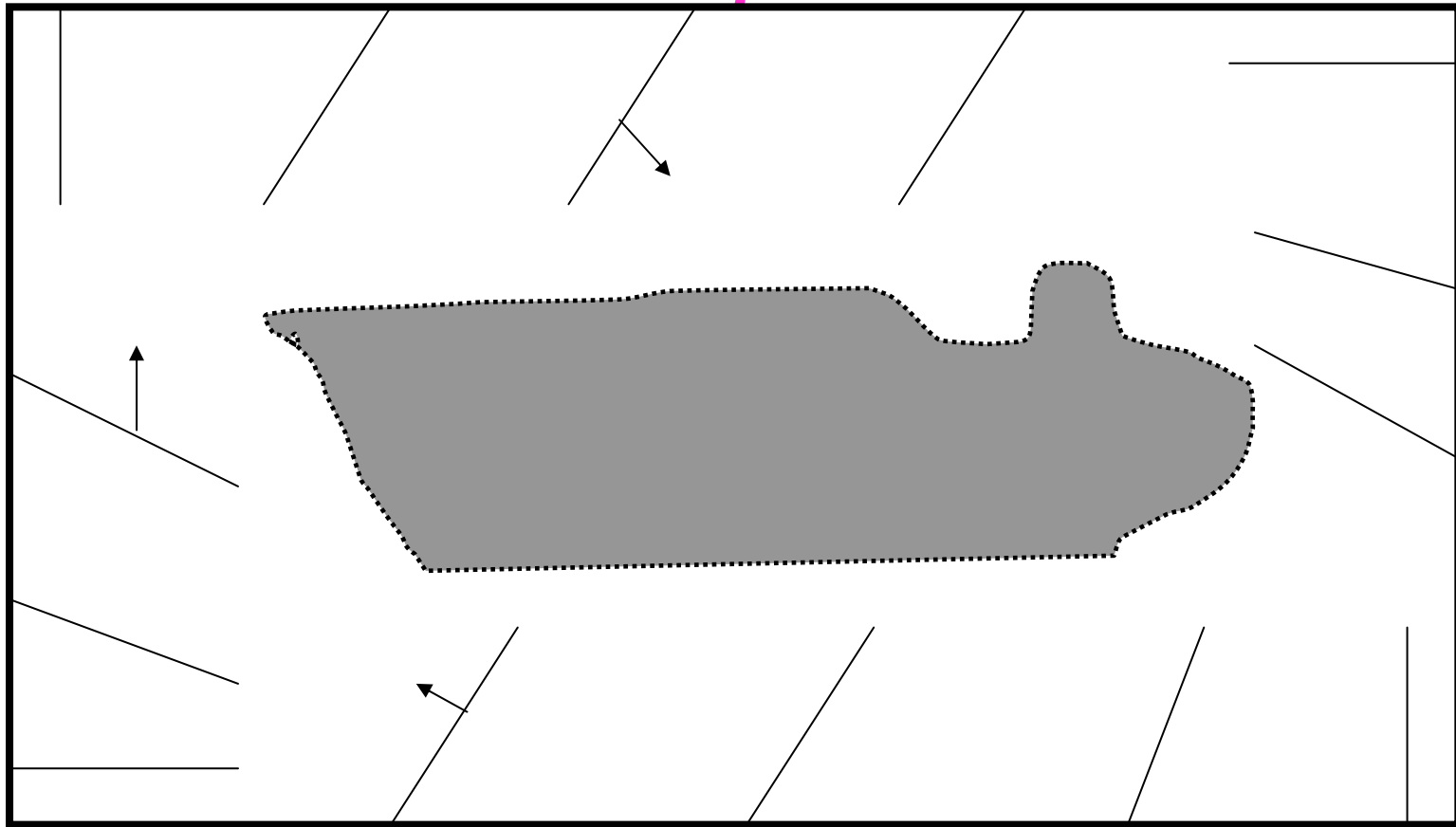


How to do it??

- 1. Enough aeration to maintain oxygen above 4-5 mg/l.
- 2. Lined pond: Plastics, concrete, soil concrete, laterite.
- 3. Placement of aerators in a way that all pond volume will be mixed.
NO ACCUMULATION OF SLUDGE!!

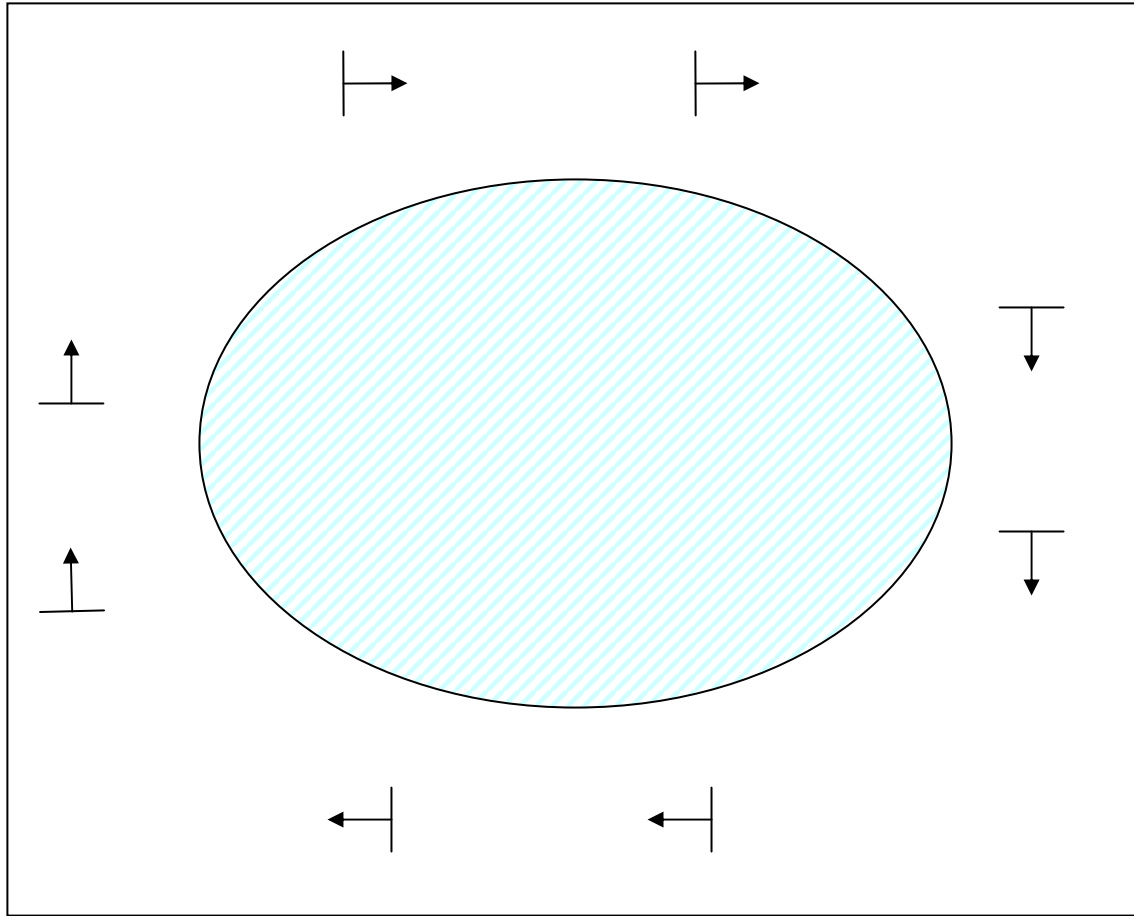


Redirecting aerators

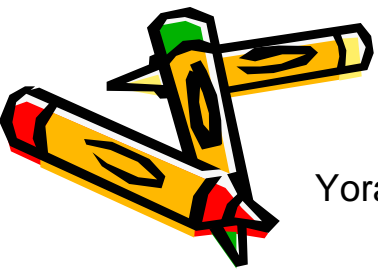
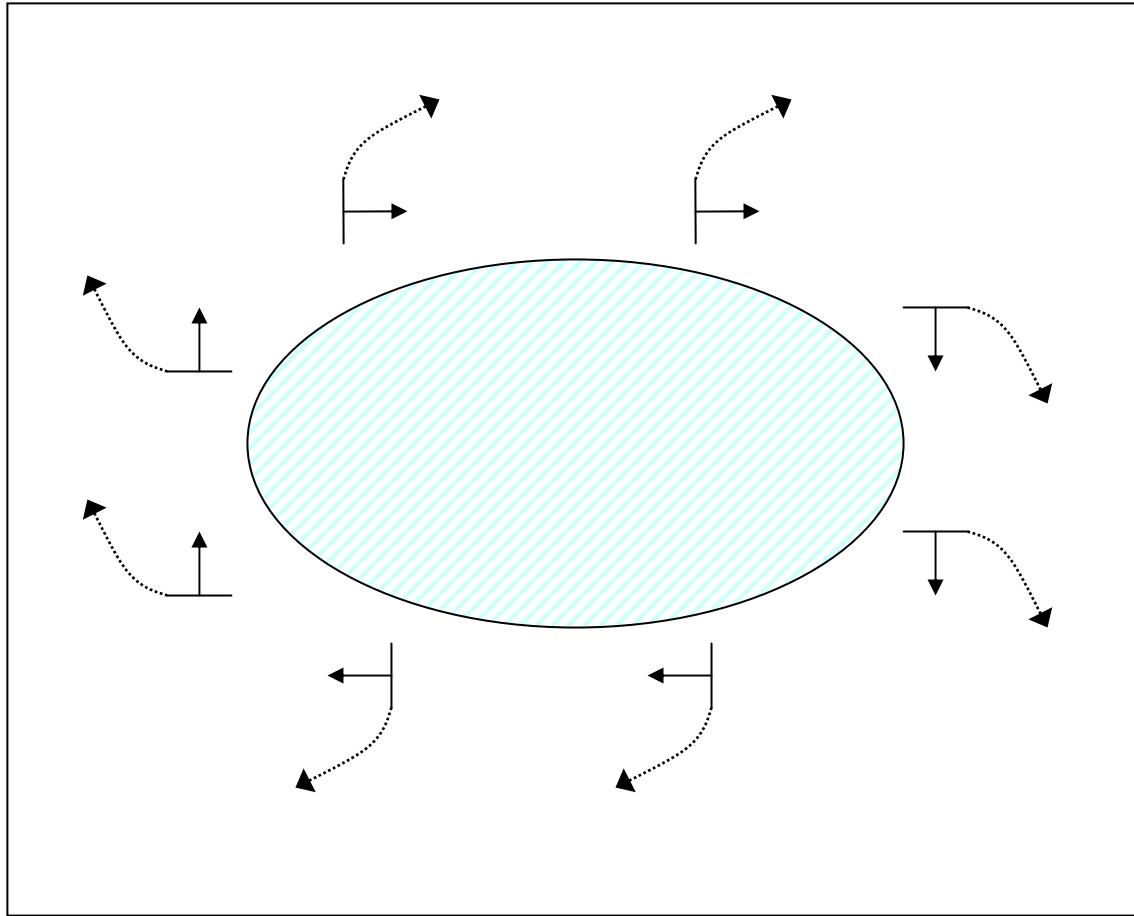


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From conventional to active

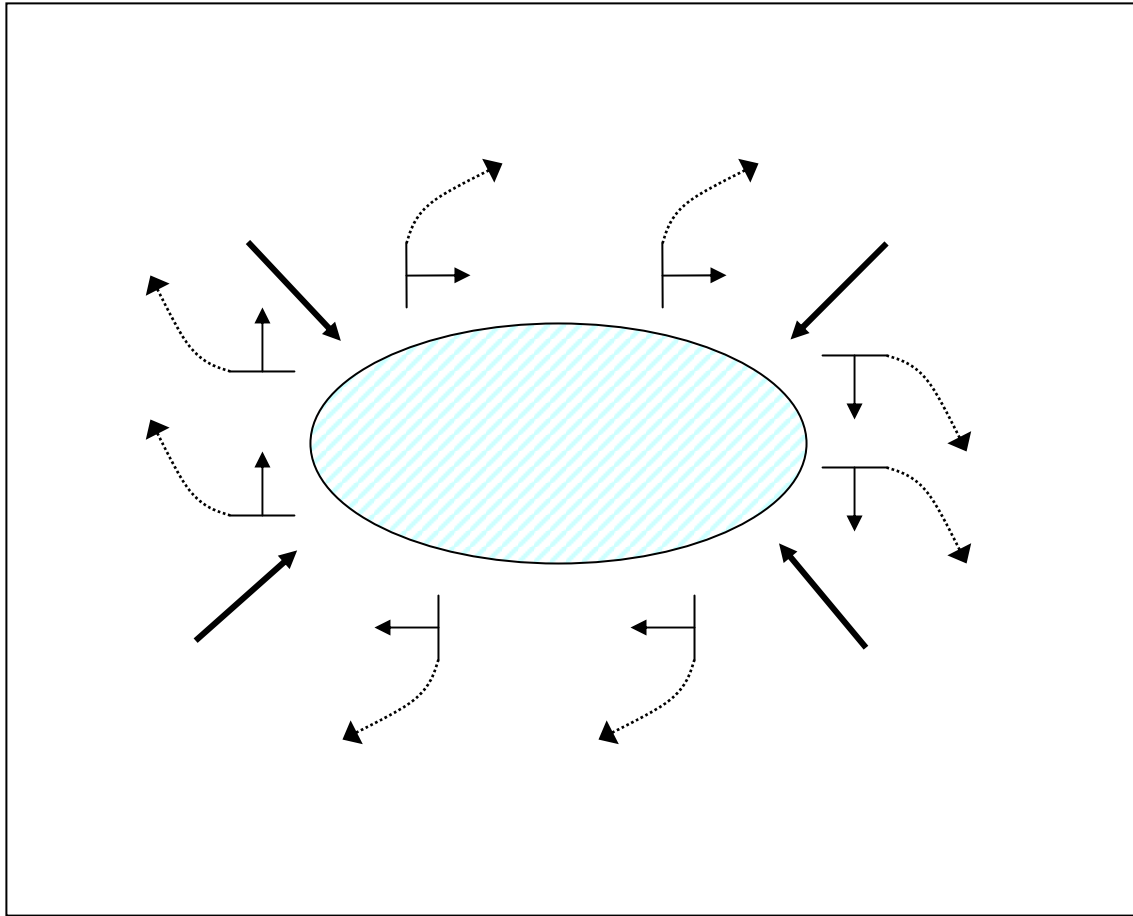


From conventional to active



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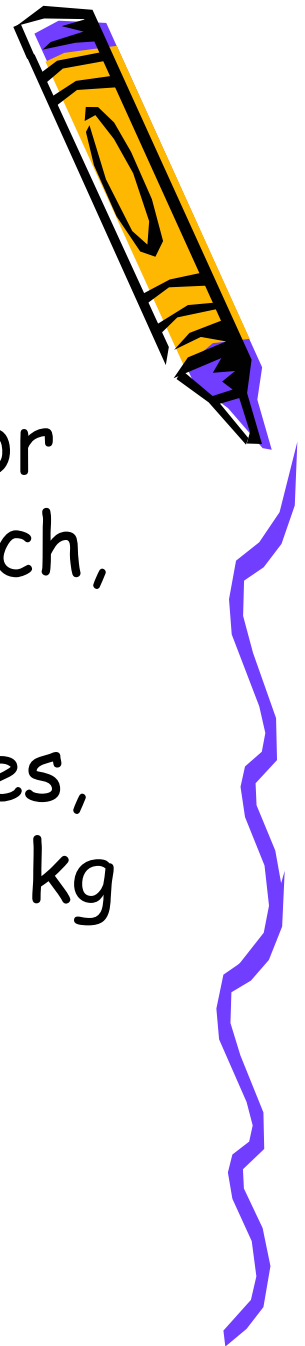
From conventional to active



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How to do it (2)

- 4. Feed with low protein % (20%) or add enough carbon (molasses, starch, cassava etc.)
- 5. If inorganic nitrogen accumulates, add carbohydrates at a rate of 20 kg per kg N you want to remove.
- 6. Maintain alkalinity > 50-100 mg.



An Example: Over-wintering of tilapia



The Problem:

- One of the problems of growing tilapia in temperate regions is over-wintering. You need the whole year to complete the growing cycle of tilapia, yet, tilapia is sensitive to cold temperature and below $\sim 13^{\circ}\text{C}$ mortality starts.
- In Israel, normally it is possible to hold juveniles in ponds without catastrophes, but if cold spells occur, mortality takes place and next year there are much less tilapia in the market.



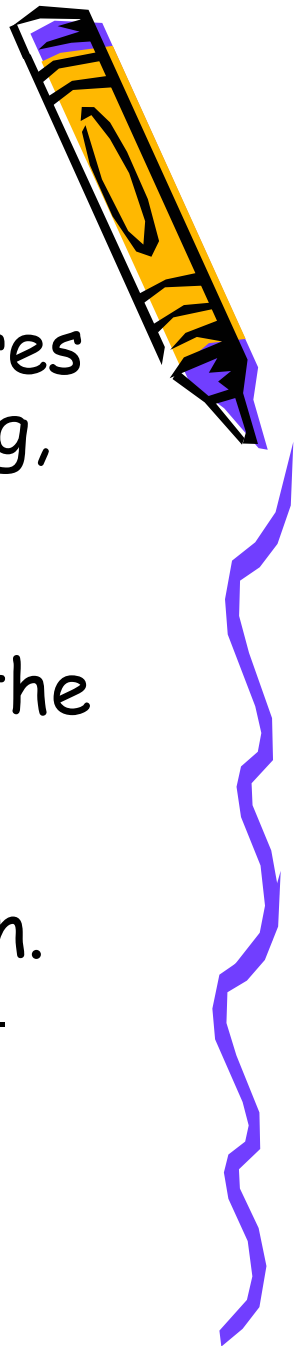
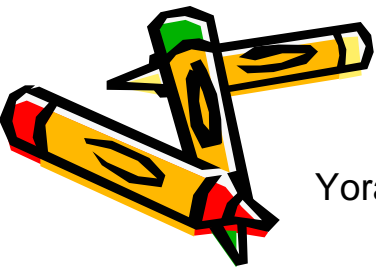
Similar situation was report from China, where production declined by ~80% due to the cold winter 2008



Potential solution

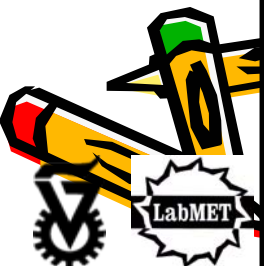
- It is possible to keep juveniles and market size fish in green houses or similar structures during the winter, usually using solar heating, but in cases geothermal water.
- These structures are expensive. A dense biomass has to be held, in order to justify the investment. In case of dense biomass, metabolites, especially ammonium may accumulate and endanger the fish population.

Water replacement is not a reasonable option, – since you loose heat by releasing the warm water.



Materials and methods

- Experimental design
 - Winter period 2008 (13th of January till 4th March)
 - 50 and 100 g tilapia hybrid fingerlings (*Oreochromis niloticus* x *Oreochromis aureus*)
 - 16 kg fish/m³ pond water (could be raised!)

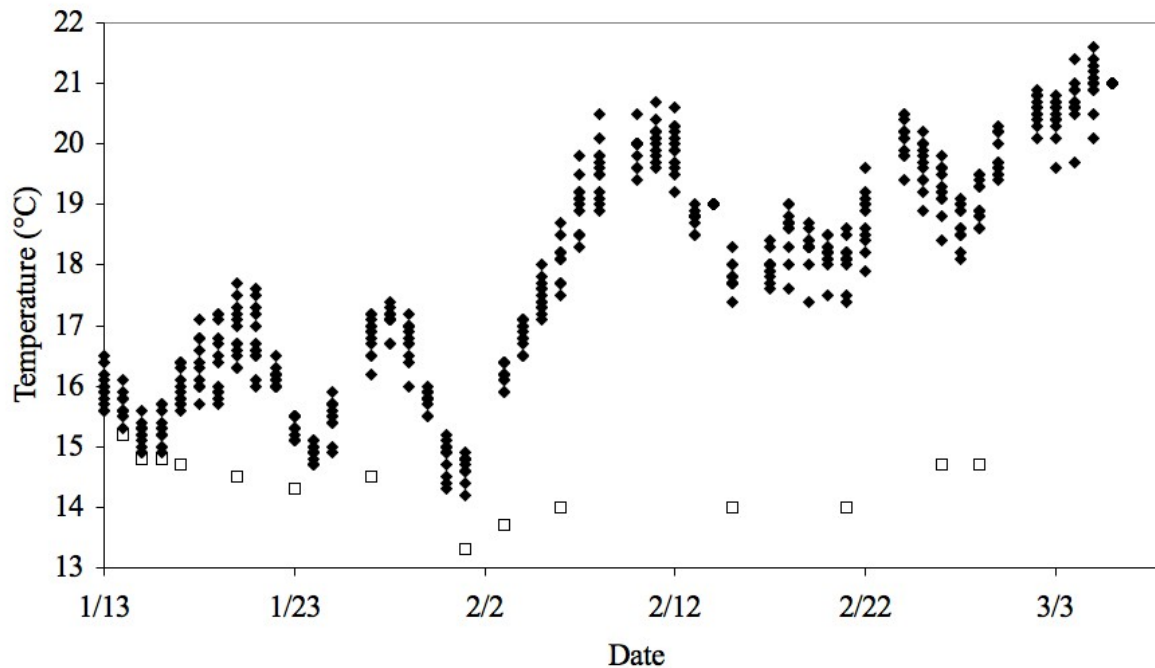


Results

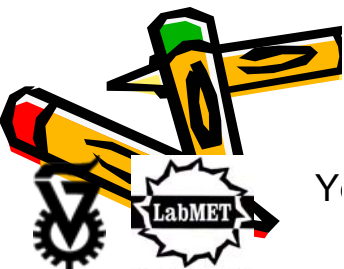
Temperature and oxygeno
control

Dissolved oxygen: 9 – 10 mg O₂/L

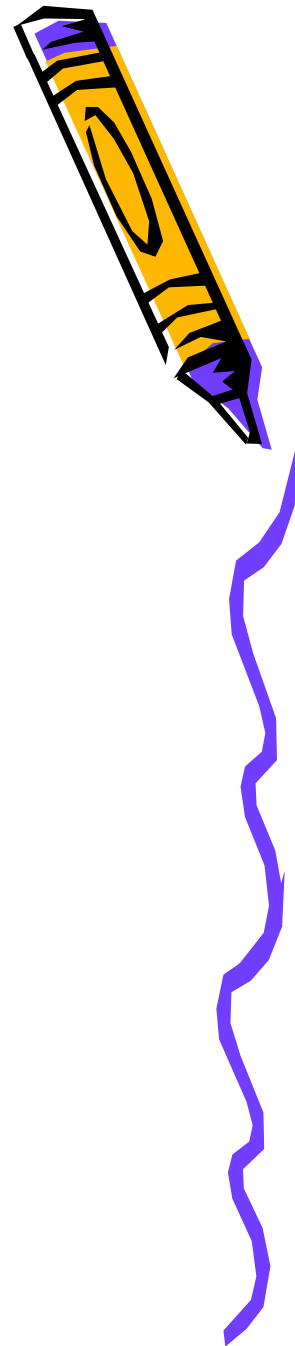
Average temperature: 18°C



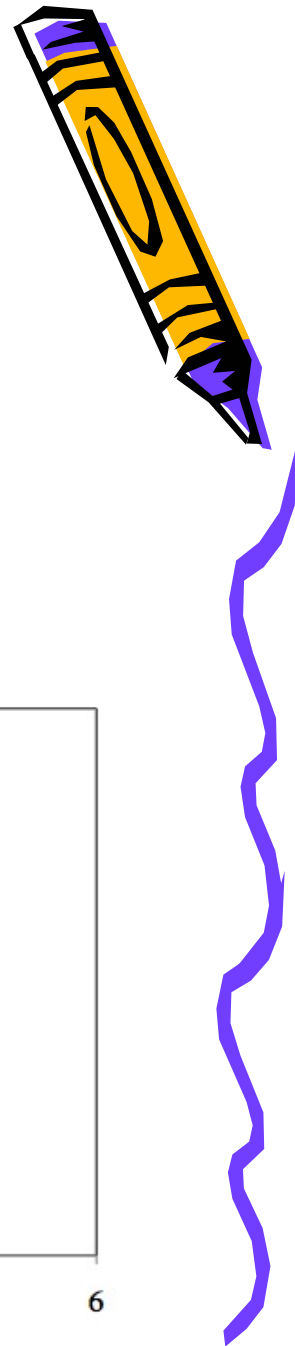
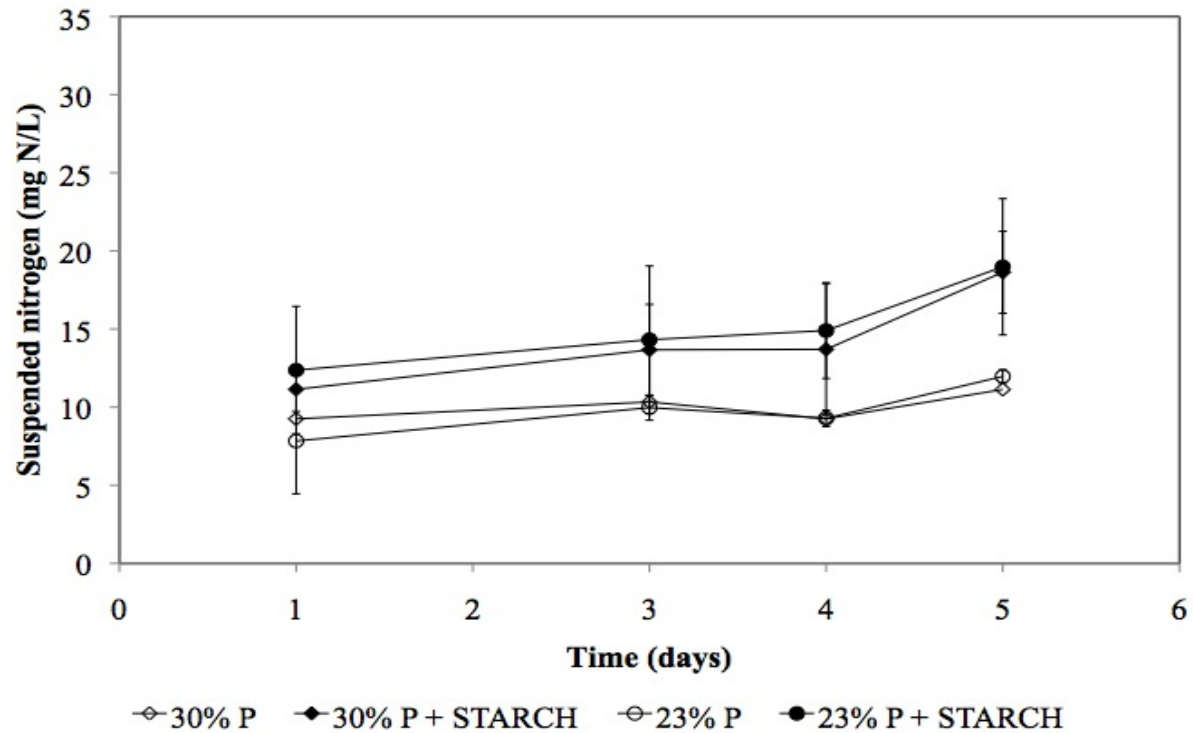
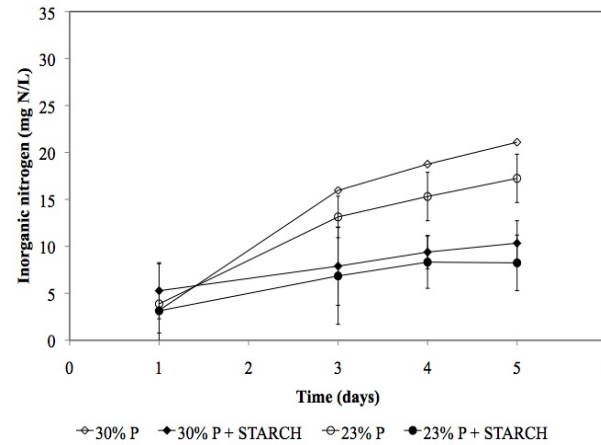
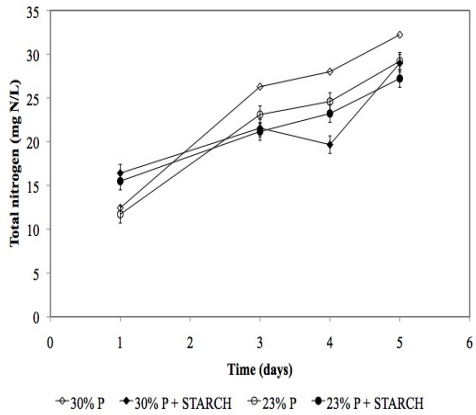
◆ Ponds □ Lake Kinneret



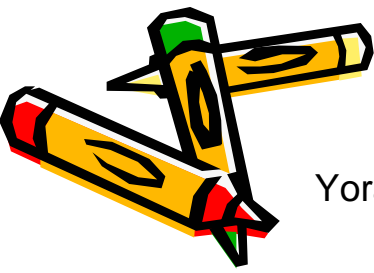
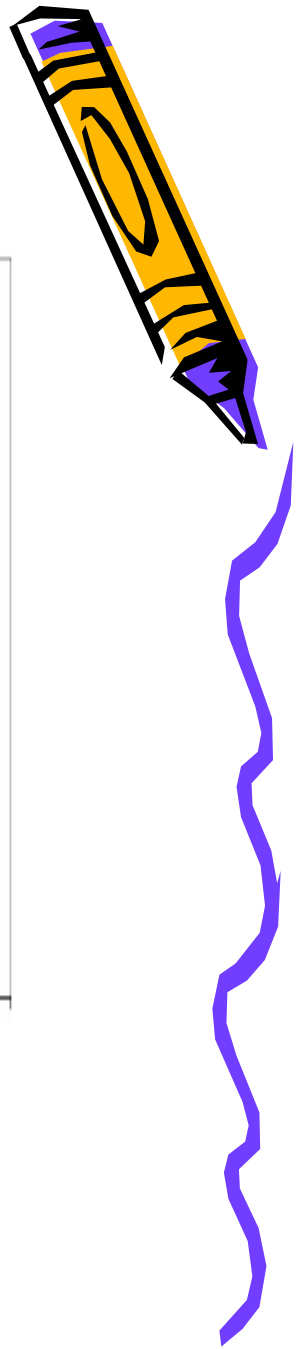
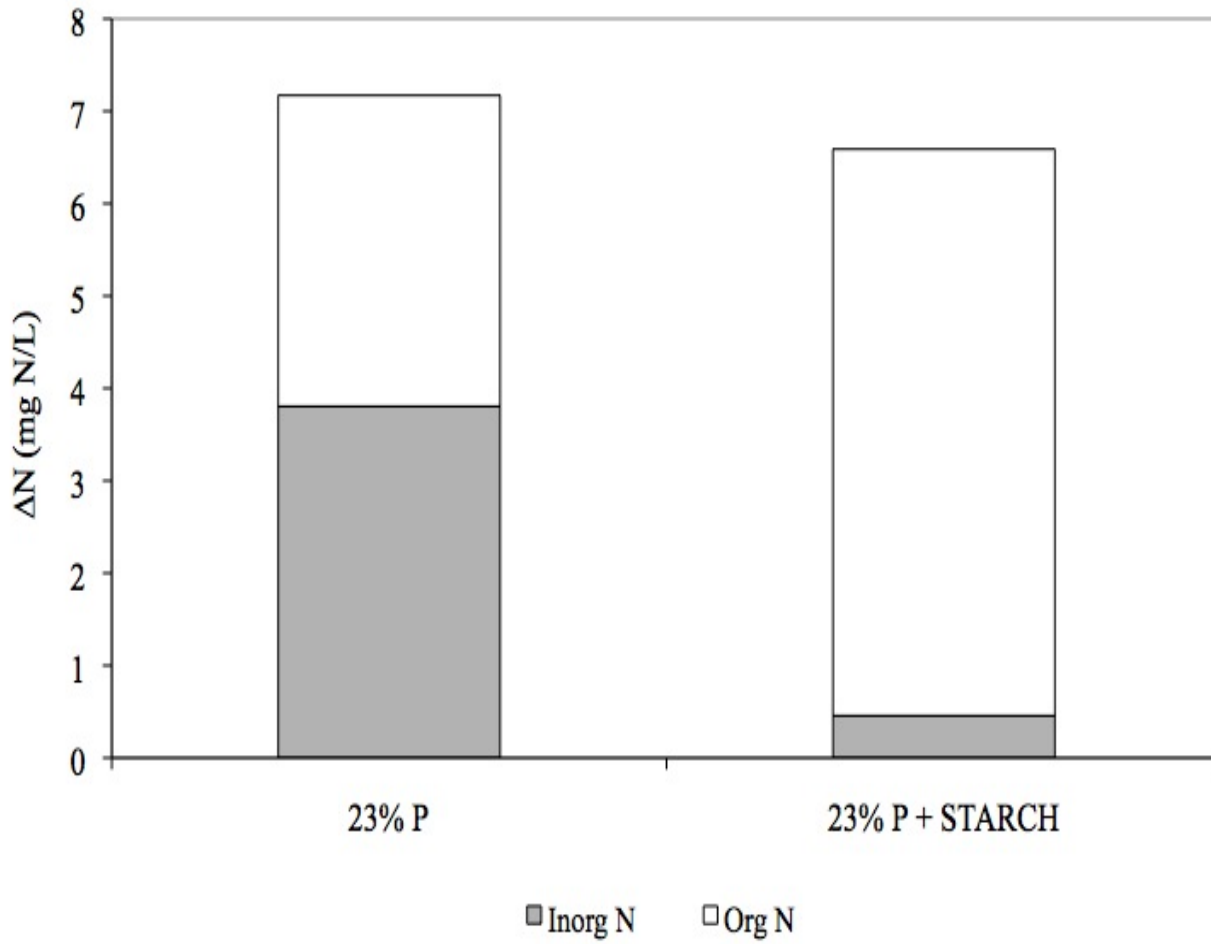
Y₁



Nitrogen dynamics



Yoram Avni



Yoram Avnimelech

Fish Responses

○ Fish survival

○ 50 g fish: $80 \pm 4\%$

○ 100 g fish: $97 \pm 6\%$

○ Fish growth

○ 50 g fish: 0.27 ± 0.02
g/fish.day

○ 100 g fish: 0.29 ± 0.03
g/fish.day

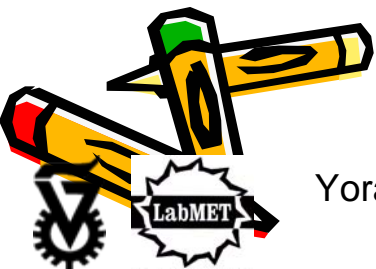
○ Condition factor, $CF = 100.W/L$

○ $CF < 1.8$ poor conditions

○ $CF > 2$ good physiological state

2.17 ± 0.06 50 g f

2.19 ± 0.07 100 g f



How to do it (3)

- 7. Minimize water exchange.
- 8. If sludge accumulates, drain sludge out or dry/clean between seasons.
- **9. It was demonstrated that the same principles can be used in stagnant ponds!!!**
- Every farm is some what different. Learn from yours and others experience.



A large floating aerator is shown in the foreground, splashing water. In the background, there are other ponds and a building with a green roof. The text is overlaid on the image in a bold, yellow font.

**We have gone a long way:
Hundreds of BFT ponds
producing shrimp and tilapia
are operative successfully all
over the globe**

- Assume 1 kg shrimp/m³
- 3% feed per day = 30g
- 40% protein = 12g
- 15.5% N in protein = 1.86 g N
- 75% is released to the water ~
- ~ 1.4 g N/m³ added daily to the water (= 1.4 ppm per day)



MICROBIAL CONVERSION

- MICROBES PRODUCE NEW CELL MATERIAL (protein) AND ENERGY:

$$\Delta C = CO_2 + \Delta C_{cell}$$

$$\Delta C_{cell}/\Delta C = \varepsilon$$

= Microbial conversion efficiency

=normally, 0.4-0.6 for aerobic microbial processes. Lower for anaerobic.

Bacteria are rich in respect to N (C:N ~ 4)

THIS, 1 Nitrogen is taken up for 4 ΔC

